

P	112es	SCIENCE	GRADE 10
Gold I	And	2nd Rank 25000 + Silver Medal + Certificate of Excellence	3rd Rank ↓ 10000 + Bronze Medal + Certificate of Excellence
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	Rs 1 + Certificate of	100 Wri	st Watch te of Excellence
Inst	tructions		
Time	: 1 hour		Maximum Marks : 100
1.	Maximum Time is 1 hour & You will the exam.	get additional ten minutes to fill up information ab	out yourself on the OMR Sheet, before the start of
2.	Write your Name, School Code, Cl	ass, Roll No. and Mobile Number clearly on the C	DMR Sheet and do not forget to sign it.
3.	The Question Paper comprises four Scientific Reasoning (25 Question (5 Questions). Each question carrie	r sections: ns), General Science (15 Questions), Logical Re a es two marks.	asoning (5 Questions) and Wise Wizard
4.	All questions are compulsory. There	e is no negative marking. Use of calculator is not pe	rmitted.

5. To mark your choice of answers by darkening the circles on the OMR Sheet, use **HB Pencil** or **Blue / Black ball point pen** only.

Roll No	
Student Name	
Father's Name	

SECTION - A (SCIENTIFIC REASONING)

- Q1. What is the magnification when the object is placed at 2*f* from the pole of a convex mirror? a. $\frac{1}{3}$ b. $-\frac{2}{3}$ c. -1 d. $-\frac{3}{2}$
- Q2. A body is moving along a circular path of radius *r*. The ratio of distance to displacement of the body, when it completes $\frac{3}{4}$ th revolution, is
 - a. $\frac{2\sqrt{2\pi}}{3}$ b. $\frac{3\pi}{2}$ c. $\frac{3\pi}{2\sqrt{2}}$ d. $\frac{\sqrt{3\pi}}{2}$
- Q3. No force acts on the charge fired through a magnetic field, when the angle between its velocity and magnetic field is

a.
$$\pi$$
 b. $\frac{3\pi}{4}$ c. $\frac{\pi}{2}$ d. $\frac{\pi}{4}$

Q4. A man is suffering from vision defect. According to given diagram, the defect he suffers from is



- a. Myopia b. Cataract c. Hypermetropia d. Astigmatism
- Q5. Work done by a body against friction always results in
 - a. Increase in kinetic energy b.
- b. Decrease in kinetic energy
 - c. Increase in potential energy d. Decrease in potential energy
- Q6. A long straight wire carries a current along *Z* axis. It is not possible to find two points in *X*-*Y* plane where
 - a. The magnetic fields are same
 - b. The direction of magnetic fields are same
 - c. The magnitude of magnetic fields are equal
 - d. The field at one point is opposite to that at the other point
- Q7. A particle travels a distance of 20 m in 7th second and 24 m in 9th second. The distance travelled by the particle in 15th second will be
 - a. 42 m b. 36 m c. 48m d. 32 m
- Q8. The weight of a person is 560 N on the surface of earth. The weight of same person at a depth
 - $\frac{R}{8}$ from the surface of earth would be
 - a. 490 N b. 560 N c. 420 N d. 360 N

Q7. Choose the mediteet statement about hydrogen as a fue	Q9.	Choose the	incorrect	statement	about l	hydrogen	as a fuel
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- a. It has the highest calorific value
- b. Its burning does not cause any pollution
- c. It can be prepared by the electrolysis of water
- d. It does not burn with explosion. So, its use is very much safe
- Q10. The main reasons for the formation of chemical bond is
 - a. To attain minimum energy and minimum stability
 - b. To attain maximum energy and maximum stability
 - c. To attain maximum energy and minimum stability
 - d. To attain minimum energy and maximum stability
- Q11. The mass of 18 molecules of water is

a. $5.38 \times 1.0-23$ g b. $5.38 \times 1.0-22$ g c. $2.98 \times 1.0-23$ g d. $5.38 \times 1.0-24$ g

- Q12. The pair of cross linked polymers is
 - a. Rayon and nylon b. Teflon and rayon
 - c. Bakelite and melamine d. PVC and PHB
- Q13. Consider the following statements :
 - I. Sodium carbonate is used in glass, soap and paper industry.
 - II. Sodium carbonate is used to remove permanent hardness of water.
 - III. Chemical formula of washing soda is NaHCO₃.

The correct statements are :

- a. I and III b. II and III c. I and II d. I, II and III
- Q14. Which of the following compounds is formed when zinc reacts with caustic soda?
 - a. Zn (OH)2 b. ZnCl2 c. ZnCO3 d. Na2ZnO2
- Q15. Consider the following box :

Mg, Al, Cu, Na, Mn, Fe, Ag

The total number of metals which react with very dilute nitric acid to give hydrogen is

- a. 3 b. 4 c. 5 d. 2
- Q16. Electronic configuration of an element 'E' is x, x^3 , $2x^2 5$. Position of element E in the Modern Periodic Table is
 - a. Period = 3 Group = 3 b. Period = 3 Group = 13
 - c. Period = 5 Group = 13 d. Period = 3 Group = 2

Q17.	All of the following are neutral salts, except				
	a. Potassium nitrate	2	b. Potassium sulphate		
	c. Potassium acetate		d. Potassium chloride		
Q18.	The process that does	s not involve ox	idation is		
	a. Alloying	b. Rusting	c. Combustion	d. Rancidity	
Q19.	Riccia belongs to the	e division			
	a. Pteridophyta	b. Bryophyta	c. Gymnospermae	d. Thallophyta	
Q20.	A cell becomes turgi	id when placed i	in		
	a. Hypertonic solut	ion	b. Hypotonic solution	n	
	c. Isotonic solution		d. Ringer's solution		
Q21.	Oxygen is consumed	l in all of the fol	llowing processes, except		

- a. Respiration b. Photosynthesis c. Decay d. Combustion
- Q22. Study the given flow chart.



Select the incorrect statement w.r.t. the cells which are indicated as letters A-F.

- a. The number of A per microlitre of blood is called TLC
- b. The letter E is used for granulocytes and squeezing out of F through the capillary wall is called diapedesis
- c. An abnormal rise in the count of B is called polycythemia
- d. The letter C is used for agranulocytes and cell indicated as D are able to squeeze out through the capillary wall

- Q23. Read the following statements and identify which of them are true and which of them are false and select the correct option.
 - (i) Branchial respiration occurs in fishes.
 - (ii) Pharynx is a common passage for the air we inhale and the food we eat.
 - (iii) In men, the vocal cords are usually shorter and thinner.
 - (iv) Breathing occurs voluntarily but the rate is controlled by the respiratory centre located in the medulla oblongata.
 - (v) Diaphragm is the characteristic of all mammals.

	(i)	(ii)	(iii)	(iv)	(v)
a.	False	False	True	True	True
b.	True	True	False	False	True
с.	True	False	True	True	False
	F 1	T	T	F 1	T

- d. False True True False True
- Q24. While walking in a garden, Mohan detected the smell of a flower. Which of the following parts firstly received the scent?
 - a. Axon of sensory neuron
 - b. Dendrite of sensory neuron
 - c. Dendrite of motor neuron
 - d. Dendrite of relay neuron
- Q25. Shaman is suffering from a disorder due to which he is excreting 8 litres of urine per day. A hormone deficiency of which of the following endocrine glands is responsible for this?
 - a. Pancreas b. Thyroid c. Pineal d. Pituitary

SECTION - B (GENERAL SCIENCE)

Q26. A ray of light is incident on a plane mirror-I and after reflection from mirror-II, enter into glass slab as shown in the figure. If refractive index of the glass slab is 2, then what is the thickness of the glass slab?



- Q27. Specific resistance of conductor depends upon
 - a. Shape of conductor b. Length of conductor
 - c. Area of conductor d. Temperature of conductor
- Q28. A long straight current carrying wire is placed near a long wire which is bent as shown in the figure. If radius of arc is 20 cm, then what will be the net magnetic field at point *P*?





Q29. The net magnetic field at the centre of a three quarter circular arc of radius *R* carrying a constant clockwise current *I*, is





Q30. In the circuit given below, A current is drawn from battery as shown in figure. Battery has internal resistance of 1 Ω thus emf of battery is



- a. 1 V b. 2 V c. 3 V d. 4 V
- Q31. Which of the following is/are alkali(es)?
 - (a) Ferric hydroxide (b) Cupric hydroxide (c) Calcium hydroxide
 - a. Only (a) b. Both (b) and (c) c. Both (a) and (c) d. Only (c)

Q32. The correct arrangement of the given elements in the decreasing order of the atomic radii is

$$a. \quad Be > Mg > Ca > Sr \\ c. \quad Sr > Ca > Mg > Be \\ d. \quad Be > Ca > Mg > Sr \\ d. \quad Ae > Ca > Mg > Sr \\ d. \quad Be > Ca > Mg > Sr \\ d. \quad Ae > Ca > Mg > Sr \\ d. \quad Ae > Ca > Mg > Sr \\ d. \quad Ae > Ca > Mg > Sr \\ d. \quad Ae > Ca > Mg > Sr \\ d. \quad Ae > Ca > Mg > Sr \\ d. \quad Ae > Ca > Mg > Sr \\ d. \quad Ae > Ca > Mg > Sr \\ d. \quad Ae > Ca > Mg > Sr \\ d. \quad Ae > Ca > Mg > Sr \\ d. \quad Ae > Ca > Mg > Sr \\ d. \quad Ae > Ca > Mg > Sr \\ d. \quad Ae > Ca > Mg > Sr \\ d. \quad Ae > Ca > Mg > Sr \\ d. \quad Ae > Ca > Mg > Sr \\ d. \quad Ae > Ca > Mg > Sr \\ d. \quad Ae > Ca > Mg > Sr \\ d. \quad Ae > Ca > Mg$$

Q33. Match the following and choose the correct option.

Column-I

Column-II

- (a) HC=C-CH=CH-CH3(i) 2- Ethyl-3-methylbut-1-ene (b) $CH_2=CH-CH-CH_2$ (ii) Pent-3-en-1-yne (c) $CH_3=CH_3$ (c) $CH_3=CH_3$ (iii) 2,2-Dimethylpropane
- (d) $\begin{array}{c} CH_3-CH-C=CH_2 \\ I \\ CH_3 \\ CH_3 \\ CH_3 \end{array}$ (iv) 3- Methylpenta-1,4-diene
- a. (a)- (iv), (b)- (ii), (c)- (iii), (d)-(i)
- b. (a)- (ii), (b)- (i), (c)- (iii), (d)- (iv)
- c. (a)- (ii), (b)- (iv), (c)-(i), (d)-(iii)
- d. a)- (ii), (b)-(iv), (c)-(iii), (d)-(i)

Q34. The common metal present in brass and bronze is

a. Lead b. Zinc c. Tin d. Copper

- Q35. A gaseous product 'X' is released by the reaction of bleaching powder with sulphuric acid. 'X' on reacting with gaseous product released at cathode during chlor-alkali process forms a compound 'Y'. Compound 'Y' is
 - a. Na2CO3 b. CaCl2 c. NaOH d. HCl
- Q36. Which of the following is a sexually transmitted disease caused by virus in human?
 - a. Syphilis b. Genital herpes c. Gonorrhoea d. Chlamydiasis
- Q37. In the given structure of nephron, identify the labelled part that is involved in the selective reabsorption of the substances and the part which is permeable to water respectively.

a. P and Sb. R and Qc. Q and Sd. Q and R



Q38. Study the given graph carefully where 'X' is a/an parameter/activity in our body.



Select the correct option w.r.t. the hormones and 'X' for which the above given graph is incorrect.

	Hormone	x
Γ	Adrenaline	Blood flow to skin
ſ	Parathyroid hormone	Blood calcium level
Γ	Insulin	Blood glucose level
	Prolactin inhibitory hormone	Secretion of prolactin

Q39. In which of the following organisms sex is not determined genetically?

a. Birds b. Fishes c. Lizards d. Humans

Q40. Carefully observe the below given sectional view of heart and choose the correct option regarding the flow of deoxygenated blood.

d. (4)



a. $E \to F \to G \to B$ b. $C \to F \to D \to A$ c. $A \to D \to F \to C$ d. $B \to G \to F \to E$

SECTION - C (LOGICAL REASONING)

- Q41. Read the given pattern carefully and answer the following question. Seven friends A, B, C, D, E, F and G are sitting in a queue such that all are facing towards north. G is sitting adjacent to E and F. D is sitting fourth to the right of C, who is sitting adjacent to A and E. No one is sitting in the left of A. Who is sitting at the 2nd position from the left end of the queue?
 a. B
 b. D
 c. C
 d. E
- Q42. In which of the following option, figure X is embedded?



Q43. In this question set of two figures X and Y showing a sequence of folding of a piece of paper. Figure Z shows the manner in which the folded paper has been cut. Choose a figure from the options which would most closely resemble the unfolded form of figure (Z).





Q44. A is father-in-law of E. M is daughter of P, who is daughter of F. C is son of E, who is father of M. Now on the basis of given information answer the following question.How is F related to E?

a. Motherb. Sisterc. Mother-in-lawd. DaughterQ45. Direction : Read the given pattern carefully and answer the following question.

▲ao9@ • e#di812%5 • rtk23!w163&@dg+ji♥

The 12th term to the left of 28th term from the left end is

a. 5 b. r c. 🛧 d. t

SECTION - D (WISE WIZARD)

Q46. Two ideal ammeters *A*, *B* and a voltmeter of a infinite resistance are connected in an electrical circuit as shown below.



If the ratio of the current flowing through ammeters is $I_A : I_B = 4 : 3$ and voltmeter reads 40 V, then the value of R_1 and R_2 will be respectively.

a. 28Ω , 21Ω if E = 100 Vb. 28Ω , 21Ω if E = 150 V

c. 21
$$\Omega$$
, 28 Ω if $E = 150$ V d. 21 Ω , 28 Ω if $E = 100$ V

Q47. Two long straight current carrying wires of current 3A and 4A are placed at the *x*-axis is shown below.

The magnitude of net magnetic field at point (5,0) is

[Take $\mu 0 = 4\pi \times 10^{-7}$ Tm/A and dimensions are in metre]

a. $2 \times 10-7$ T b. $3 \times 10-7$ T c. $4 \times 10-7$ T d. $5 \times 10-7$ T

Q48. Consider the following,

$$A + Na_2CO_3 \longrightarrow B + H_2O + CO_2 \uparrow$$

$$A + NaHCO_3 \longrightarrow CH_3COONa + H_2O + Y \uparrow$$

$$Y + Ca(OH)_2 \longrightarrow Z + H_2O$$

- Compounds 'Z' and 'A' respectively are a. CaO, CH₃COOH
- c. CaCO₃,CH₃COOH

b. CaO, CH_3CH_2OH

d. CaCO₃,CH₃CH₂OH

Q49. P- It represents the two alternative forms of a gene.

Q - It is the total number of genes present in the haploid set of chromosomes.

R – The type of sex determination found in grass hoppers.

S- It is the change in gene frequency which occurs by chance in a small population.

Select the option which correctly identifies P–S.

	Р	Q	R	S		
(1)	Allele	Genome	Protenor type	Genetic drift		
(2)	Factor	Genome	XX-XO type	Gene flow		
(3)	Factor	Heterosome	ZW-ZZ type	Gene pool		
(4)	Allele	Autosome	Lygaeus type	Genetic drift		
a	. (1)		b. (2)		c. (3)	d.(4)

Q50. Refer the given dichotomous key and select the correct option for P, Q, R and S (I) (a) It is a peptide hormone - Go to (II)

(b) It is a steroid hormone - Go to (IV)

(II) (a) It is produced from pancreas - Go to (III)(a)

(b) It is produced from posterior pituitary - Go to (III) (b)

(III) (a) It stimulates glucose production by liver - \mathbf{P}

(b) It promotes water reabsorption by the kidneys - ${f Q}$

(IV) (a) It is secreted from adrenal gland Go to (V) (a)

(b) It is secreted from gonads - Go to (V)(b)

(V) (a) It stimulates gluconeogenesis - **R**

(b) It regulates activity in female reproductive tissues – \mathbf{S}

57. 1	Р	Q	R	S
(1)	Glucagon	Vasopressin	Cortisol	Estrogen
(2)	Thyroxine	Vasopressin	Corticotropin	Aldosterone
(3)	Insulin	Glucagon	Progesterone	Epinephrine
(4)	Cortisol	Corticotropin	Adrenaline	Progesterone

a. (1)	b. (2)	c. (3)	d (4)
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